

Biofriendly nanocomposite containers with inhibition properties for the protection of metallic surfaces

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Abstract

© 2017 The Author(s) Published by the Royal Society. All rights reserved. An attempt to combine two 'green' compounds in nanocomposite microcontainers in order to increase protection properties of waterborne acryl-styrene copolymer (ASC) coatings has been made. N-lauroylsarcosine (NLS) served as a corrosion inhibitor, and linseed oil (LO) as a carrier-forming component. LO is compatible with this copolymer and can impart to the coating self-healing properties. For the evaluation of the protective performance, three types of coatings were compared. In the first two, NLS was introduced in the coating formulation in the forms of free powder and micro-containers filled with LO, correspondingly. The last one was a standard ASC coating without inhibitor at all. Low-carbon steel substrates were coated by these formulations by spraying and subjected subsequently to the neutral salt spray test according to DIN ISO 9227. Results of these tests as well as the data obtained by electrochemical study suggest that such containers can be used for the improvement of adhesion of ASC-based coatings to the substrate and for the enhancement of their protective performance upon integrity damage, whereas the barrier properties of intact coatings were decreased.

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Keywords

Corrosion protection, Encapsulation, N-lauroylsarcosine, Waterborne acryl-styrene copolymer

References

- [1] Gupta SDV. 1993 Encapsulated corrosion inhibitors. In Reviews on corrosion inhibitor science and technology (eds A Raman and P Labine), Houston, TX: NACE.
- [2] Joshua Du Y, Damron M, Tang G, Zheng H, Chu C-J, Osborne JH. 2001 Inorganic/organic hybrid coatings for aircraft aluminum alloy substrates. *Prog. Org. Coat.* 41, 226-232. (doi:10.1016/s0300-9440(01)00133-3)
- [3] Regulation E.C. 1999 No 1907/2006 of the European Parliament and of the Council of 18 December 2006, concerning the Registration, Evaluation, Authorization and Restriction of Chemicals (REACH).
- [4] Alawadhi K, Brown R, Alsarraf J. 2014 A replacement of chromate coating for adhesive bonding of stainless steel in a corrosive environment. *IJRET* 3, 396. (doi:10.14419/ijet.v3i3.3041)
- [5] DeBerry DW. 1986 Inhibition of pitting of type 304 L stainless steel by N-lauroylsarcosine. *J. Electrochem. Soc.* 133, 30-37. (doi:10.1149/1.2108536)
- [6] Wei Z, Duby P, Somasundaran P. 2005 Inhibition of pitting corrosion by surfactants as a function of temperature. *Corrosion* 61, 341-347. (doi:10.5006/1.3279886)

- [7] Li Y, Zhang Y, Jungwirth S, Seely N, Fang Y, Shi X. 2014 Corrosion inhibitors for metals in maintenance equipment: introduction and recent developments. *Corros. Rev.* 32, 163-181. (doi:10.1515/corrrev-2014-0002)
- [8] Badran BM, Mohammed HA, Aglan HA. 2002 Effect of different polymers on the efficiency of water-borne methyl amine adduct as corrosion inhibitor for surface coatings. *J. Appl. Polym. Sci.* 85, 879-885. (doi:10.1002/app.10704)
- [9] Bai Z, Wen Z, Wiltshire JC. 2008 Marine mineral tailings use in anticorrosive coatings. *OCEANS 2008 (IEEE)*, pp. 1-9. (doi:10.1109/OCEANS.2008.5151930)
- [10] Li J, Ecco L, Delmas G, Whitehouse N, Collins P, Deflorian F, Pan J. 2015 In-situ AFM and EIS study of waterborne acrylic latex coatings for corrosion protection of carbon steel. *J. Electrochem. Soc.* 162, pC55-CC63. (doi:10.1149/2.0851501jes)
- [11] Shangguan LJ, Ma W, Liu G, Song W. 2015 Anticorrosive properties of styrene-acrylic resins containing aluminum tripolyphosphate. *Chemik* 69, 578-585.
- [12] Salensky GA, Cobb MG, Everhart DS. 1986 Corrosion-inhibitor orientation on steel. *Ind. Eng. Chem. Proc. Res. Dev.* 25, 133-140. (doi:10.1021/i300022a002)
- [13] He X, Shi X. 2009 Self-repairing coating for corrosion protection of aluminum alloys. *Prog. Org. Coat.* 65, 37-43. (doi:10.1016/j.porgcoat.2008.09.003)
- [14] Shchukin DG. 2013 Container-based multifunctional self-healing polymer coatings. *Polym. Chem.* 4, 4871-4877. (doi:10.1039/C3PY00082F)
- [15] Grigoriev D, Akcakayiran D, Schenderlein M, Shchukin D. 2014 Protective organic coatings with anticorrosive and other feedback-active features: micro- and nanocontainers-based approach. *Corrosion* 70, 446-463. (doi:10.5006/0976)
- [16] Grigoriev D. 2014 Anticorrosion coatings with self-recovering ability based on damage-triggered micro- and nanocontainers. In *Intelligent coatings for corrosion control*, ch. 8 (eds A Tiwari, L Hihara, J Rawlins). London, UK: Butterworth-Heinemann.
- [17] Wei H et al. 2015 Advanced micro/nanocapsules for self-healing smart anticorrosion coatings. *J. Mater. Chem. A* 3, 469-480. (doi: 10.1039/C4TA04791E)
- [18] Szabó T, Telegdi J, Nyikos L. 2015 Linseed oil-filled microcapsules containing drier and corrosion inhibitor-their effects on self-healing capability of paints. *Prog. Org. Coat.* 84, 136-142. (doi:10.1016/j.porgcoat.2015.02.020)
- [19] Jadhav RS, Hundiwal DG, Mahulikar PP. 2011 Synthesis and characterization of phenol-formaldehyde microcapsules containing linseed oil and its use in epoxy for self-healing and anticorrosive coating. *J. Appl. Polym. Sci.* 119, 2911-2916. (doi:10.1002/app.33010)
- [20] De Viguerie L, Payard PA, Portero E, Walter P, Cotte M. 2016 The drying of linseed oil investigated by Fourier transform infrared spectroscopy: historical recipes and influence of lead compounds. *Prog. Org. Coat.* 93, 46-60. (doi:10.1016/j.porgcoat.2015.12.010)
- [21] Mallégol J, Lemaire J, Gardette J-L. 2000 Drier influence on the curing of linseed oil. *Prog. Org. Coat.* 39, 107-113. (doi:10.1016/S0300-9440(00)00126-0)
- [22] Stenberg C, Svensson M, Johansson M. 2005 A study of the drying of linseed oils with different fatty acid patterns using RTIR-spectroscopy and chemiluminescence (CL). *Ind. Crop. Prod.* 21, 263-272. (doi:10.1016/j.indcrop.2004.04.002)
- [23] Lazzari M, Chiantore O. 1999 Drying and oxidative degradation of linseed oil. *Polym. Degrad. Stab.* 65, 303-313. (doi:10.1016/S0141-3910(99)00020-8)
- [24] Rammelt U, Reinhard G. 1992 Application of electrochemical impedance spectroscopy (EIS) for characterizing the corrosion-protective performance of organic coatings on metals. *Prog. Org. Coat.* 21, 205-226. (doi: 10.1016/0033-0655(92)87005-U)
- [25] Itou T, Kitai H, Shimazu A, Miyazaki T, Tashiro K. 2014 Clarification of cross-linkage structure in boric acid doped poly(vinyl alcohol) and its model compound as studied by an organized combination of X-ray single-crystal structure analysis, Raman spectroscopy, and density functional theoretical calculation. *J. Phys. Chem. B.* 118, 6032-6037. (doi:10.1021/jp5026569)
- [26] Shibayama M, Sato M, Kimura Y, Fujiwara H, Nomura S. 1988 ¹¹B NMR study on the reaction of poly(vinyl alcohol) with boric acid. *Polymer* 29, 336-340. (doi:10.1016/0032-3861(88)90343-6)
- [27] Yin Y, Li J, Liu Y, Li Z. 2005 Starch crosslinked with poly(vinyl alcohol) by boric acid. *J. Appl. Polym. Sci.* 96, 1394-1397. (doi:10.1002/app.21569)
- [28] Nass KK. 1988 Representation of the solubility behavior of amino acids in water. *AIChE J.* 34, 1257-1266. (doi:10.1002/aic.690340804)
- [29] Borisova D, Akcakayiran D, Schenderlein M, Möhwald H, Shchukin DG. 2013 Nanocontainer-based anticorrosive coatings: effect of the container size on the self-healing performance. *Adv. Funct. Mater.* 23, 3799-3812. (doi:10.1002/adfm.201203715)
- [30] Ji G, Ouyang Z, Li G, Ibekwe S, Pang S-S. 2010 Effects of adhesive thickness on global and local mode-I interfacial fracture of bonded joints. *Int. J. Solids Struct.* 47, 2445-2458. (doi:10.1016/j.jisols.2010.05.006)